Former date: 21st September 2006

1.	IDENTIFICATION OF THE CHEMICAL AND OF THE MANUFACTURER, IMPORTER OR OTHER UNDERTAKING				
1.1		on of the substance or prepar	ration		
	Trade name				
	ice Power in	nstant Cold compress			
1.2	Use of the chemical				
1.2.1	The intended use				
	Cold theraph	•			
1.3	Identification of the manufacturer, importer or other undertaking				
1.3.1	Manufacturer, importer, other undertaking Fysioline Oy Pharma				
1.3.2	Contact info				
	Street addre	ess	Arv	ionkatu 2	
		nd post office	338	40 Tampere, Finland	
	E-mail	•		@fysioline.fi	
	Telephone r	number		8 32330 300	
	Telefax		+35	8 32330 333	
	Y code		FIC	927611-8	
1.3.3	Information	on foreign manufacturer			
1.4	Emergency	telephone			
	PL 340, Haa	tric of Helsinki and uusimaa, In ırtmanninkatu 4 Finland	toxication Informat	ion Central (24h)	
2.	PL 340, Haa 00029 HUS,	artmanninkatu 4		ion Central (24h)	
2. 2.1	PL 340, Haa 00029 HUS,	artmanninkatu 4 Finland ION AND INFORMATION ON		ion Central (24h)	
2.1 2.1.1 CAS	PL 340, Haa 00029 HUS, COMPOSITI Hazardous i number or	ITTMANNINKATU 4 Finland ION AND INFORMATION ON ingredients 2.1.2 Name of the	INGREDIENTS	2.1.4 Warning symbol, R phrases and	
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5.2	Water Extinguishing media which must not be used for safety reasons
5.3	Special exposure hazards in a fire
5.4	Special protective equipment for fire-fighters
5.5	Other instructions
6.	ACCIDENTAL RELEASE MEASURES
6.1	Personal precautions
6.2	Environmental precautions
6.3	Methods for cleaning up Wash with water.
6.4	Other instructions Small amounts of ammonium nitrate solid can be collected in a clean and dry container.
7.	HANDLING AND STORAGE
7.1	Handling
7.2	Storage Stored in room temperature.
7.3	Specific use(s)
8.	EXPOSURE CONTROLS/PERSONAL PROTECTION
8.1	Exposure limit values
8.1.1	HTP values
8.1.2	Other limit values
8.1.3	Limit values in other countries
8.2	Exposure controls
8.2.1	Occupational exposure controls
8.2.1.1	Respiratory protection
8.2.1.2	Avoid breathing ammonium nitrate dust if the pack is broken. Hand protection
8.2.1.3	Eye protection
8.2.1.4	Skin protection Use clothing or a towel between skin and cold compress to avoid freezing.
8.2.2	Environmental exposure controls
9.	PHYSICAL AND CHEMICAL PROPERTIES
9.1	General information (physical state, colour and odour)
	Inodorous, granules and water bag inside white and blue plastic bag pack.
9.2	Important health, safety and environmental information
9.2.1	pH Ammoniumnitrate inside the package: 5.4 (0.1M water suspension).
9.2.2	Boiling point/boiling range Ammoniumnitrate inside the package: 210 celsius (break down).
9.2.3	Flash point
9.2.4	Flammability (solid, gas)

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9.2.5	Explosive properties		
9.2.5.1	Lower explosive limit		
9.2.5.2	Upper explosive limit		
9.2.6	Oxidising properties Ammoniumnitrate inside the package: Strong oxidating agent.		
9.2.7	Vapour pressure		
9.2.8	Relative density		
9.2.9	Solubility		
9.2.9.1	Water solubility		
	Ammoniumnitrate inside the package: good water soluble (2 500 g/l).		
9.2.9.2	Fat solubility (solvent-oil to be specified) Ammoniumnitrate inside the package: Dissolves to the acetone, ethanol (38 g/l), chloroform, methanol (171 g/l) and bases. Indissolve to the ethers.		
9.2.10	Partition coefficient: n-octanol/water		
9.2.11	Ammonium nitrate inside the pack: $P_{ow}=0.8 \times 10^{-3}$, log $P_{ow}=-3.1$, no affiny for lipids. Viscosity		
9.2.12	 Vapour density 		
9.2.13	Evaporation rate		
9.3	Other information		
10.	STABILITY AND REACTIVITY		
10.1	Conditions to avoid		
10.2	Materials to avoid Ammoniumnitrate inside the pack:Combustible substances, agents, acids, alkali, sulphur, chlorates, chlorides, chromates, nitrites, permanganates, metal powders and substances containing copper, nickel,		
10.3	cobalt, zinc and alloys of any aforementioned materials. Hazardous decomposition products Ammoniumnitrate inside the pack: In contact with alkaline metals: ammonia. Intensiveheating in closed premises: may cause active reaction or explosion.		
11.	TOXICOLOGICAL INFORMATION		
11.1	Acute toxicity		
11.2	 Irritation and corrosiveness Ammoniumnitrate inside the package might irritate eyes and skin.		
11.3	Sensitisation		
11.4	Sub-acute, sub-chronic and prolonged toxicity		
11.5	Empirical data on effects on humans		
11.6	Other information on health effects		
12.	ECOLOGICAL INFORMATION		
12.1	Ecotoxicity		
12.1.1	Aquatic toxicity Ammoiniumnitrate inside the pack: Toxicity to the aquatic organisms depends on waters pH.		
	Ammoniumnitrate is harmfull for aquatic organisms.		
12.1.2			

Completely soluble in water.

12.3	Persistence and degradability
12.3.1	Biodegradation
	Ammoniumnitrate is biologically disintegrable.
12.3.2	Chemical degradation
12.4	Bioaccumulative potential
	Not accumulative in the foodchain.
12.5	Other adverse effects
10	
13.	DISPOSAL CONSIDERATIONS
	Can be disposed together with domestic waste. Ammoniumnitrate can be used by fertilizer if the package has been broken.
14.	TRANSPORT INFORMATION
14.1	UN number
14.0	 Decking group
14.2	Packing group
14.3	Land transport
14.4	Sea transport
14.5	Air transport
15	
15. 15.1	REGULATORY INFORMATION Information on the warning label
15.1.1	Letter code of the warning symbol and indications of danger for the preparation
15.1.2	Names of the ingredients given on the warning label
45 4 0	 Durburger
15.1.3	R phrases
15.1.4	S phrases
16.	OTHER INFORMATION
16.1	List of the relevant R phrases
16.2	 Training advice
16.3	Restrictions on use
16.4	Further information
16.5	Sources of key data used
	Material safety data sheet delivered by raw material manufacturer.

16.6 Information which has been added, deleted or revised